
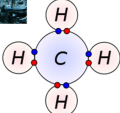
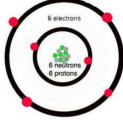

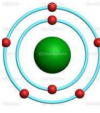

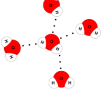
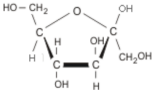


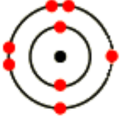


1. Group the following according to type.
 2. Determine what each picture represents within the group or type.
 3. Describe what each is showing and define or identify it specifically.


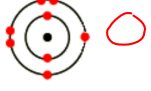

TYPES OF BOND **TYPES OF ATOMS**
PHASES OF MATTER **ISOMERS**

● Electron from hydrogen
 ● Electron from carbon

TYPES OF ATOMS

masses Proton - p⁺
 Neutrons - no charge N
 electrons - e⁻
 #e⁻ = #p⁺
 mass = #p⁺ + #N

Na⁺ *Cl⁻*

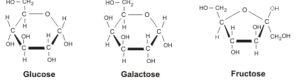
Ionic Bond - btwn 2 ions

Na 11p⁺ 10e⁻ → Na⁺
Cl 17p⁺ 18e⁻ → Cl⁻

*C has 4e⁻ in ring wants 8
 H only 1e⁻ wants 2e⁻ to fill*

Covalent Bond - share a pair of e⁻

ISOMERS



Glucose Galactose Fructose

Figure 3: Glucose, galactose et fructose représentés en HAWORTH